### SEMINARIUM RENTGENOWSKIE

### Seminarium w trybie stacjonarnym

### Laboratorium SL-1 zaprasza na seminarium, które odbędzie się dnia 15.04.2025 r. o godz. 10:30 w Sali D

### <u>Tytuł Seminarium</u>:

# High-pressure chemistry and physics of fluorides

Current diamond anvil cell (DAC) techniques enable performing routine measurements on solids compressed to 100 GPa (million atmospheres). At such conditions the pressure-volume work term (pV) becomes comparable to covalent bond energies therefore considerably affecting the chemistry, structure, and properties of compounds that are otherwise well known and characterized under ambient conditions. [1,2]

Through a combination of Raman spectroscopy and synchrotron powder X-ray diffraction measurements supplemented with density functional theory (DFT) calculations we unraveled the pressure response of two fluoride systems. The first was K<sub>2</sub>CuF<sub>4</sub>, a prototypical 2D ferromagnetic system. We found that at 10 sliding of  $[CuF_4]^{2-}$  layers leads to a transition GPa from the Ruddlesden-Popper phase into a Dion-Jacobson-like structure. This transition results in substantial structural and electronic rearrangement within the planes. resulting in а change from 2D ferromagnetism to 1D antiferromagnetism. [3]

The second system is palladium trifluoride which despite it seemingly simple stoichiometry is a mixed-valent system better formulated as  $Pd^{II}Pd^{IV}F_6$ . We performed an attempt to verify whether application of high pressure might force this compound to form a genuine  $Pd^{III}$  fluoride  $(Pd^{III}F_3)$ . Indeed, hybrid density functional calculations predict the thermodynamic preference for single-valent (comproportionated) polymorphs at pressures exceeding 30 GPa. Experimentally we found two phase transitions, with the second one, commencing at ~50 GPa, introducing a monoclinic *C2/c* phase containing genuine  $Pd^{III}$  centers. Preliminary data suggests that the this phase might host strong one-dimensional antiferromagnetic spin-spin interactions.

#### References

W. Grochala, R. Hoffmann, J. Feng, N. W. Ashcroft, *Angew. Chemie Int. Ed.* 46, 3620, 2007.
L. Zhang, Y. Wang, J. Lv, Y. Ma, *Nat. Rev. Mater.* 2, 17005.

[3] S. B. Pillai, D. Upadhyay, J. Drapała, Z. Mazej, D. Kurzydłowski, J. Phys. Chem. C 128, 17747, 2024.

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