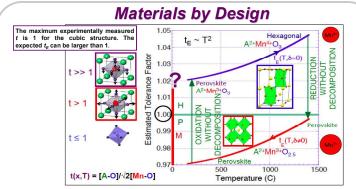
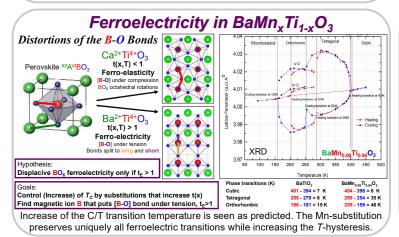
Synthesis of the Mn-ion Based Perovskite Multiferroics

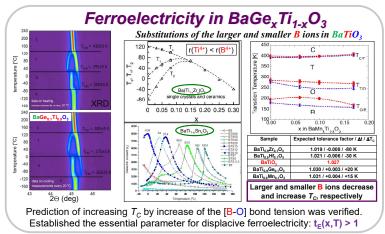
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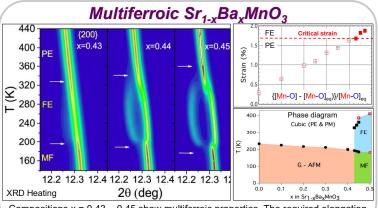
We have designed and synthesized manganese multiferroics Sr_{1-x}Ba_xMn_{1-v}Ti_vO₃ exhibiting ferroelectricity (FE) and magnetism originating solely from the Mn⁴⁺ (d³) ion. The structural study show large FE-type distortions, which are reduced by antiferromagnetism indicating strong coupling. The Ti-substituted compounds have increased transition temperatures and larger derived FE polarizations than BaTiO₃. The scarcity of similar multiferroic systems is caused by the difficulty of obtaining compounds with expected tolerance factor $t_{F} > 1$ near the room temperature, and not due to the incompatibility of the magnetic systems with the FE (d⁰) as suggested by the "d⁰-ness conjecture" (Hill, J. Phys. Chem. B, 2000)



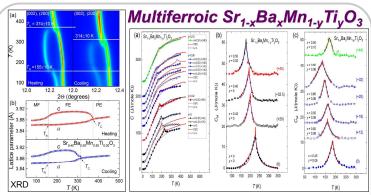
Two-step synthesis was used to remove hexagonal non-FE phase by: Reducing the oxygen content at high temperatures to form perovskite framework by decreasing the tolerance factor below putting back oxygen at low temperature by annealing.







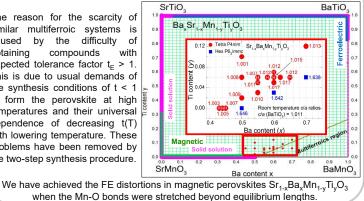
Compositions x = 0.43 - 0.45 show multiferroic properties. The required elongation of the [Mn-O] bond is 1.7% to split it into long and short bonds along the c-axis



The FE transitions show huge temperature hysteresis at $T_{\rm C}$, an increased c/a ratio and T_C to 420 K, as well as reduced suppression of distortion at T_N . Specific heat studies show that AFM transition changes from the continuous to discontinuous



The reason for the scarcity similar multiferroic systems is caused the difficultv bv of with obtaining compounds expected tolerance factor $t_E > 1$. This is due to usual demands of the synthesis conditions of t < 1 to form the perovskite at high temperatures and their universal dependence of decreasing t(T)with lowering temperature. These problems have been removed by the two-step synthesis procedure.



The two-step synthesis, which uses chemical pressure instead of hydrostatic pressure provides a method to synthesize the large size samples of novel compounds for multivalent transition and rare earth metals with varying ionic sizes. We have used the method to stabilize as well as to destabilize several perovskites and non-perovskites for basic studies and application.

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